

## **Foldamers**

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## Fine Tuning of β-Peptide Foldamers: a Single Atom Replacement Holds Back the Switch from an 8-Helix to a 12-Helix

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Abstract: Cyclic homologated amino acids are important building blocks for the construction of helical foldamers. Naminoazetidine-2-carboxylic acid (AAzC), an aza analogue of trans-2-aminocyclobutanecarboxylic acid (tACBC), displays a strong hydrazino turn conformational feature, which is proposed to act as an 8-helix primer. tACBC oligomers bearing a single N-terminal AAzC residue were studied to evaluate the ability of AAzC to induce and support an 8-helix along the oligopeptide length. While tACBC homooligomers assume a dominant 12-helix conformation, the aza-primed oligomers preferentially adopt a stabilized 8-helix conformation for an oligomer length up to 6 residues. The (formal) single-atom exchange at the N terminus of a tACBC oligomer thus contributes to the sustainability of the 8-helix, which resists the switch to a 12-helix. This effect illustrates atomic-level programmable design for fine tuning of peptide foldamer architectures.

Coldamers are unnatural oligomers that adopt well-defined folded patterns, and several factors that govern their conformational preferences have been identified.<sup>[1]</sup> β-Peptides were amongst the first foldamers to be studied and they provide benchmark helix-forming manifolds, [2,3] the diversity of which has been enlarged through the study of  $\alpha/\beta$ -,  $\beta/\gamma$ -, and other mixed peptides.<sup>[3,4]</sup> Oligomers of cyclic β-amino acids adopt stabilized helices where the pitch depends largely on the backbone torsional angle  $\theta$  (N-C<sup> $\beta$ </sup>-C<sup> $\alpha$ </sup>-C(=O)), which in turn is determined by the ring size and its stereochemistry. Robust 14-helix<sup>[5,6]</sup> and 12-helix<sup>[7]</sup> foldamers can be constructed rationally from appropriate cyclic trans-β-amino acids, while the alternating 10/12-helix[8] and regular 6ribbon strands<sup>[9]</sup> are observed for oligomers of cyclic cis-βamino acids. Nonetheless, helical folding may be influenced by other factors, such as side-chain interactions, [10] the steric bulk of the monomer, [11] solvent/concentration effects, [12] or the oligomeric environment into which the monomer is placed.[13] Further exploration of such modulating factors could therefore enable the fine tuning of programmable helical folding.

Oligomers of trans-2-aminocyclobutanecarboxylic acid (tACBC) can adopt a 12-helix conformation both in solution and in the solid state.<sup>[14]</sup> However, the preferred conformer of a tACBC dipeptide is an 8-membered hydrogen-bonded ring (C8),<sup>[15]</sup> and it has been suggested that a tACBC tetrapeptide might display three consecutive C8 structural features.<sup>[16]</sup> This corresponds to an 8-helix, which is a rarity for β-peptides.<sup>[17,18]</sup> The 12-helix preference of tACBC oligomers raises the question of the sustainability of an 8-helix (Figure 1);

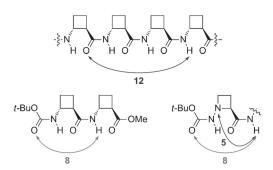


Figure 1. 8- vs. 12-membered H-bonded rings in tACBC oligomers and the Hz-turn conformation of an AAzC residue.

indeed, in previous cases where 12-helix conformers compete, the 8-helix was only observed for small (<4-mer) oligomers.<sup>[17]</sup> N-Aminoazetidine-2-carboxylic acid (AAzC)<sup>[19]</sup> is an aza analogue of tACBC and is characterized by a strong tendency to form a bifurcated C8/5-ring H-bonded structure known as a hydrazino (Hz) turn (Figure 1). [20,21] We therefore examined the ability of a single AAzC residue to behave as an "8-helix primer" when employed as the N-terminal residue in an oligo-tACBC sequence.

Four oligopeptides (1–4; Figure 2) were prepared (see the Supporting Information) in order to compare their conformational behavior with that of the corresponding homooligomers Boc(tACBC) $_n$ OMe (5–8; n = 2, 4, 6, 8, respectively). [14] The two series are identical with the exception of the formal

Figure 2. Structures of peptides 1-4.

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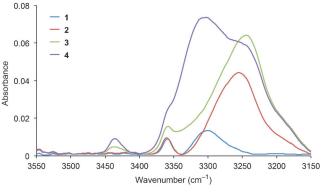
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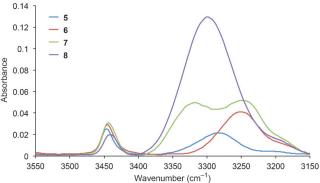


Figure 3. IR absorption spectra (in CHCl<sub>3</sub>) of AAzC peptides 1-4 (top) and tACBC oligomers 5-8 (bottom).

single-atom exchange (N for C) at the  $\beta$ -position of the N-terminal residue in the former series.

In CHCl<sub>3</sub> solution, the N–H stretch absorptions in the IR spectra were useful for tracking conformational changes with increasing peptide length in the two series of peptides (Figure 3). In the reference *t*ACBC oligomer series, **5** showed an H-bond absorption band at 3290 cm<sup>-1</sup> consistent with a C8 conformer. This absorption appeared with progressive red-shift in **6** and **7**, although the spectrum of the latter showed the presence of a second H-bonded conformer. This second conformer was predominant in the spectrum of **8**. These data are consistent with a predominant 8-helix conformation for **6**, a 12-helix conformer for **8**, and the coexistence

of these two conformers for **7**. In the AAzC peptide series **1**–**4**, the diagnostic small N–H absorbance at 3360 cm<sup>-1</sup> indicated the presence of an Hz-turn in all cases. Peptides **1**–**3** showed a strong H-bond absorbance, which became progressively red-shifted from 3300 to 3240 cm<sup>-1</sup> with increasing peptide length. The spectrum of **4** featured an additional strong H-bond absorption at 3300 cm<sup>-1</sup> and an absorption corresponding to free N–H at 3440 cm<sup>-1</sup>. Comparison of the two series clearly suggests that in CHCl<sub>3</sub> solution, the 12-helix is already prevalent in the *t*ACBC hexamer and predominant in the octamer, while for the AAzC peptides, an Hz-turn-promoted 8-helix is predominant up to the hexapeptide and remains a significant conformer for the octapeptide; the second conformer is not clearly identified at this stage.

These observations were corroborated by NMR analysis of the AAzC peptides 2-4 in CDCl<sub>3</sub> solution. In each case, the N<sup>4</sup>H <sup>1</sup>H NMR signal appeared at  $\delta \approx 6.5$  ppm and a group of strongly deshielded signals at  $\delta \approx 9.0$  ppm was observed for the other amide groups, thus suggesting the latter to be involved in strong hydrogen bonds. The intramolecular nature of the H-bonds in 2 and 3 was confirmed by the absence of a significant variation in chemical shift with concentration, which is consistent with an 8-helix structure. For 4, only the  $N^{10}H$  signal ( $\delta = 8.99$  ppm) was strongly deshielded and was uninfluenced by concentration, thus suggesting the presence of an Hz-turn. Other amide signals were moderately deshielded ( $\delta = 7.86-8.70 \text{ ppm}$ ) and showed moderate concentration dependence, thus suggesting the presence of both H-bonded and non-H-bonded conformations, which is in agreement with a conformational equilibrium for 4. The N<sup>4</sup>H signal was an exception, with significant concentration and isotopic exchange effects suggesting that it remains essentially non-H-bonded in 4.

Further evidence for the H-bond networks in peptides **2–4** in CDCl<sub>3</sub> was acquired by using fast-pulsing high-field SOFAST HMBC experiments. The diagnostic  $N^{10}H\cdots N^{5}$  interaction of an Hz-turn was ascertained for all three peptides by  $^{1}H^{N_{-}15}N$  SOAFST HMBC. A  $^{1}H^{N_{-}13}CO$  SOFAST HMBC experiment for **3** revealed a series of ( $i\rightarrow i-2$ ) H-bonds, which together with the Hz-turn, constitute an uninterrupted 8-helix (Figure 4). This is entirely consistent

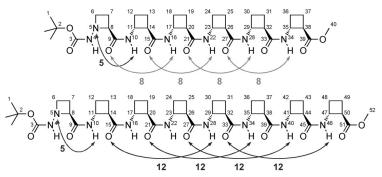


Figure 4. Hydrogen bonding in peptides 3 (in CDCl $_3$ ) and 4 (in [D $_5$ ]pyridine) observed by SOFAST HMBC NMR experiments.

with the IR data. A CDCl<sub>3</sub> solution of 4 failed to give usable 2D NMR data, probably owing to the coexistence of two interconverting conformers. NMR studies were carried out in  $[D_5]$ pyridine solution, thus allowing the identification of the second conformer of 4. The Hz-turn was confirmed by  ${}^{1}\text{H}^{-15}\text{N}$  SOFAST HMBC, while  ${}^{1}\text{H}^{-13}\text{CO}$  SOFAST HMBC revealed four  $(i \rightarrow i - 3)$  H-bonds that define a 12-helical segment involving all seven *tA*CBC residues (Figure 4). Peptide 4 thus adopts a unique hybrid conformation in pyridine, starting with an 8-membered Hz-turn and switching to a 12-helix for the rest of the sequence. This conformer coexists with the 8-helix for a solution of 4 in CHCl<sub>3</sub>.

The low solubility of peptides 1-4 in protic solvents ( $<5 \, \text{mm}$  in  $\text{CD}_3\text{OH}$ ) precluded similar NMR studies in such media. However, the CD spectra were recorded in dilute



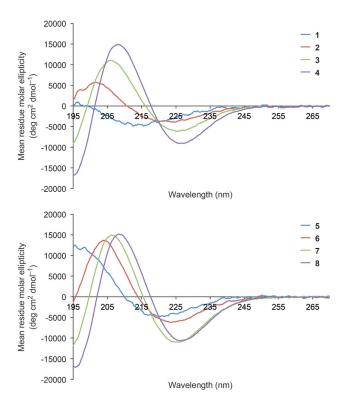


Figure 5. CD spectra (in MeOH) of AAzC peptides 1-4 (top) and tACBC oligomers 5-8 (bottom).

MeOH solution (0.2 mm) and compared with those of the tACBC oligomers 5-8 (Figure 5). Comparison of AAzC peptides 3 and 4 with oligomers 6 and 7, respectively, is revealing. While the contribution from a 12-helix (max. 207 nm, min. 227 nm)<sup>[14]</sup> is already significant in tetramer 6, this structure is only partially in evidence for hexapeptide 3. Tetrapeptide 2 displayed a different curve (max. 203 nm, min. 223 nm): prevalence of an 8-helix conformation is an attractive hypothesis but this should be treated with caution since the CD signal of the  $\beta$ -peptide 8-helix has not been calculated. All CD curve intensities were only partially diminished over the temperature range 5–55 °C, with no detectable conformer change.[23]

A hybrid MCMM conformational search was carried out on peptides 2-4 in vacuum by using MacroModel 10.6<sup>[24]</sup> and the MMFF94s force field without restraints. Low-energy families were subjected to ab initio geometrical optimization at the B3LYP 6-31G\*\* level of theory in CHCl<sub>3</sub> and in MeOH using GAUSSIAN 09. [25] The Hz-turn of AAzC dominated in low-energy conformers, regardless of the other conformational features. Only two low-energy conformer families emerged: the 8-helix and the Hz-turn/12-helix hybrid (Figure 6). In CHCl<sub>3</sub>, the 8-helix was the most stable conformer for 2 and 3, whereas for 4, the hybrid conformer was only 0.7 kJ mol<sup>-1</sup> higher in energy. These results are in complete agreement with the experimental data, thus suggesting the predominance of the 8-helical conformer up to six residues, and the presence of two conformers for octapeptide 4. In MeOH, the 8-helix is the lowest energy conformer for tetrapeptide 2 by at least 8.6 kJ mol<sup>-1</sup>. This would explain the low CD ellipticity of 2, which can now be considered more

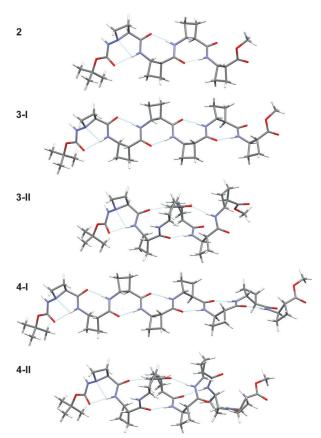


Figure 6. Calculated low-energy conformers of peptides 2-4.

confidently as the signature of an 8-helix. For 3, the 8-helix (3-I) and the hybrid Hz-turn/12-helix (3-II) have closer energies and for 4, the hybrid structure 4-II is more stable. These observations account qualitatively for the increasing "12helix-like" contribution to CD data with increasing peptide

Inspection of the backbone torsional angles  $(\phi, \theta, \psi)$ revealed the remarkable simplicity of the helix pitch switch: the AAzC residues in 4-I and 4-II are virtually superimposable, with a  $\psi$  value (ca. 4.0°) that is amenable to both 8-helix and 12-helix progressions. In each helix type, the tACBC residues have similar values for  $\phi$  (range 87° to 100°) and  $\theta$ (range  $-95^{\circ}$  to  $-103^{\circ}$ ); only the  $\psi$  values differ significantly (ca. 35° in 4-II, ca. 95° in 4-II).

A single crystal of 4 was obtained through slow evaporation of a CH<sub>2</sub>Cl<sub>2</sub>/MeOH solution and was analyzed by X-ray diffraction (Figure 7).[26] The main characteristics of the hybrid Hz-turn/12-helix conformer 4-II were in evidence: the N-terminal AAzC shows the three-centered H-bonded Hz-turn, while the tACBC segment adopted a 12-helix structure into which a molecule of methanol had been inserted between N<sup>34</sup>H (residue 3) and O=C<sup>21</sup> (residue 6), with minor distortion of the helical axis.[27]

From these studies, it emerges that replacement of the Nterminal residue of a tACBC oligomer by AAzC, conceptually a single-atom substitution, has the effect of enhancing 8-helical folding prompted by the strong Hz-turn of AAzC. The 8-helices described herein are the longest described to date for β-amino acid oligomers. In CHCl<sub>3</sub>, the 8-helix is the



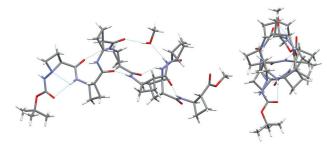


Figure 7. X-ray diffraction structure of peptide 4: side view (left) and view along the axis (right).

main conformer up to an octapeptide, whereas the corresponding tACBC oligomers largely exist in 12-helical conformations.<sup>[14]</sup> In MeOH, the 8-helix makes a significant contribution up to a 6-residue sequence when primed by AAzC. When the 8-helix prompt is not followed, the tACBC segment folds into a 12-helix, giving a hybrid Hz-turn/12-helix conformer. These results contribute to our understanding of the subtle factors that may govern the preferences for 8- or 12helix folding, and this understanding could be employed advantageously to the design of polymorphic molecular architectures.

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- [1] a) Foldamers: Structure, Properties, and Applications (Eds.: S. Hecht, I. Huc), Wiley-VCH, Weinheim, 2007; b) C. M. Goodman, S. Choi, S. Shandler, W. F. DeGrado, Nat. Chem. Biol. 2007, 3, 252-262; c) A. Roy, P. Prabhakaran, P. K. Baruah, G. J. Sanjayan, Chem. Commun. 2011, 47, 11593-11611; d) G. Guichard, I. Huc, Chem. Commun. 2011, 47, 5933-5941.
- [2] a) D. Seebach, J. Gardiner, Acc. Chem. Res. 2008, 41, 1366-1375; b) S. H. Gellman, Acc. Chem. Res. 1998, 31, 173-180.
- [3] P. G. Vasudev, S. Chatterjee, N. Shamala, P. Balaram, Chem. Rev. **2011**. 111. 657 – 687.
- [4] a) L. K. A. Pilsl, O. Reiser, Amino Acids 2011, 41, 709-718; b) A. D. Bautista, C. J. Craig, E. A. Harker, A. Schepartz, Curr. Opin. Chem. Biol. 2007, 11, 685-692; c) M.-I. Aguilar, A. W. Purcell, R. Devi, R. Lew, J. Rossjohn, A. I. Smith, P. Perlmutter, Org. Biomol. Chem. 2007, 5, 2884-2890; d) M. Salwiczek, E. K. Nyakatura, U. I. M. Gerling, S. Ye, B. Koksch, Chem. Soc. Rev. **2012**, 41, 2135 – 2171; e) W. S. Horne, S. H. Gellman, Acc. Chem. Res. 2008, 41, 1399-1408.
- [5] D. H. Appella, L. A. Christianson, I. L. Karle, D. R. Powell, S. H. Gellman, J. Am. Chem. Soc. 1996, 118, 13071-13072.
- [6] D. Seebach, M. Overhand, F. N. M. Kühnle, B. Martinoni, Helv. Chim. Acta 1996, 79, 913-941.

- [7] D. H. Appella, L. A. Christianson, D. A. Klein, D. R. Powell, X. Huang, J. J. Barchi, Jr., S. H. Gellman, Nature 1997, 387, 381 -
- [8] a) D. Seebach, K. Gademann, J. V. Schreiber, J. L. Matthews, T. Hintermann, B. Jaun, L. Oberer, U. Hommel, H. Widmer, Helv. Chim. Acta 1997, 80, 2033-2038; b) T. A. Martinek, I. M. Mándity, L. Fülöp, G. K. Tóth, E. Vass, M. Hollósi, E. Forró, F. Fülöp, J. Am. Chem. Soc. 2006, 128, 13539-13544.
- [9] a) T. A. Martinek, G. K. Tóth, E. Vass, M. Hollósi, F. Fülöp, Angew. Chem. Int. Ed. 2002, 41, 1718-1721; Angew. Chem. 2002, 114, 1794 - 1797; b) F. Rúa, S. Boussert, T. Parella, I. Díez-Pérez, V. Branchadell, E. Giralt, R. M. Ortuño, Org. Lett. 2007, 9, 3643 - 3645.
- [10] a) I. M. Mándity, A. Monsignori, F. Fülöp, E. Forró, F. Fülöp, Chem. Eur. J. 2014, 20, 4591-4597; b) P. I. Arvidsson, M. Rueping, D. Seebach, Chem. Commun. 2001, 649-650.
- [11] A. Hetényi, Z. Szakonyi, I. M. Mándity, É. Szolnoki, G. K. Tóth, T. A. Martinek, F. Fülöp, Chem. Commun. 2009, 177 – 179.
- [12] É. Szolnoki, A. Hetényi, T. A. Martinek, Z. Szakonyi, F. Fülöp, Org. Biomol. Chem. 2012, 10, 255-259.
- [13] I. M. Mándity, E. Wéber, T. A. Martinek, G. Olajos, G. K. Tóth, E. Vass, F. Fülöp, Angew. Chem. Int. Ed. 2009, 48, 2171-2175; Angew. Chem. 2009, 121, 2205-2209.
- [14] C. Fernandes, S. Faure, E. Pereira, V. Théry, V. Declerck, R. Guillot, D. J. Aitken, Org. Lett. 2010, 12, 3606-3609.
- [15] E. Torres, E. Gorrea, E. Da Silva, P. Nolis, V. Branchadell, R. M. Ortuño, Org. Lett. 2009, 11, 2301 - 2304.
- [16] E. Gorrea, G. Pohl, P. Nolis, S. Celis, K. K. Burusco, V. Branchadell, A. Perczel, R. M. Ortuño, J. Org. Chem. 2012, 77,
- [17] a) R. Threlfall, A. Davies, N. M. Howarth, J. Fisher, R. Cosstick, Chem. Commun. 2008, 585-587; b) K. Gademann, A. Hane, M. Rueping, B. Jaun, D. Seebach, Angew. Chem. Int. Ed. 2003, 42, 1534-1537; Angew. Chem. 2003, 115, 1573-1575; c) R. J. Doerksen, B. Chen, J. Yuan, J. D. Winkler, M. L. Klein, Chem. Commun. 2003, 2534-2535.
- [18] a) J. D. Winkler, E. L. Piatnitski, J. Mehlmann, J. Kasparec, P. H. Axelsen, Angew. Chem. Int. Ed. 2001, 40, 743-745; Angew. Chem. 2001, 113, 765-767; b) S. Abele, P. Seiler, D. Seebach, Helv. Chim. Acta 1999, 82, 1559-1571.
- [19] V. Declerck, D. J. Aitken, J. Org. Chem. 2011, 76, 708-711.
- [20] A. Altmayer-Henzien, V. Declerck, D. Merlet, J.-P. Baltaze, J. Farjon, R. Guillot, D. J. Aitken, J. Org. Chem. 2013, 78, 6031-6039
- [21] "Synthesis and Chemistry of  $\alpha$ -Hydrazino Acids": J. Vidal in Origins and Synthesis of Amino Acids, Vol. 2 (Ed.: A. B. Hugues), Wiley-VC, Weinheim, 2009, pp. 35-92.
- [22] A. Altmayer-Henzien, V. Declerck, D. J. Aitken, E. Lescop, D. Merlet, J. Farjon, Org. Biomol. Chem. 2013, 11, 7611-7615.
- [23] The CD spectra of 3 and 4 could be taken in MeOH:H<sub>2</sub>O (1:1) with no apparent change.
- [24] MacroModel, version 10.6, Schrödinger, LLC New York, NY, 2014.
- [25] Gaussian 09, Revision D.01, M. J. Frisch, et al., Gaussian, Inc. Wallingford CT, 2009.
- [26] CCDC 1051739 (4) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.
- [27] A crystal of 3 was grown under similar conditions. The crystal quality was poorer, but X-ray diffraction at 1.1 Å resolution showed the structure was that of the hybrid conformation 3-II; see the Supporting Information.

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